PY-403(O&N)

B. Pharm. (Fourth Semester) EXAMINATION, June, 2011

PHARMACEUTICAL ANALYSIS-I

[PY-403(O&N)]

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Time: Three Hours

Maximum Marks: 70

Note: Attempt any *five* questions. All questions carry equal marks. It is required to solve subparts of a question in continuation.

- 1. Discuss various analytical techniques employed in Pharmaceutical industries and add a note on significance of quantitative analysis in quality control.
- 2. Explain the neutralization curve for the following titrations:
 - (a) Strong acid vs Strong base
 - (b) Weak acid vs Strong base
- 3. (a) What are redox titrations? How will you balance the redox reactions?
 - (b) Give principle and procedure of permanganate titration.
- 4. What is argentometric titration? Discuss the endpoint determination methods of precipitation titration.

- 5. Give the principles of gravimetric analysis. Discuss briefly the gravimetric analytical techniques.
- 6. What are the solvents used in non-aqueous titrations?

 Discuss the non-aqueous titrations of weak acids and weak bases with suitable examples.
- 7. What is conductometry titration? Discuss the principle and applications of conductometric titrations with suitable examples.
- 8. Write short notes on any three of the following: $4\frac{1}{2}$ each
 - (a) Karl-Fisher titration
 - (b) Principles of polarography
 - (c) Factors influencing the stability of metal complexes
 - (d) Iodometry titration

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